

BIOSYNTHESIS OF FATTY ACIDS FROM [CARBOXY-¹⁴C] ACETATE BY SOLUBLE ENZYME SYSTEM PREPARED FROM RAT MAMMARY GLAND HOMOGENATES

by

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We have described the biosynthesis of short- and long-chain fatty acids (C_6 – C_{18}) from acetate by homogenates prepared from the mammary gland of lactating rats and of sheep¹. In the preparations made from the rat tissue three conditions were required for the activation of the fatty-acid synthesizing system: (a) aerobic incubation; (b) the addition of either of the three keto-acids, pyruvate, oxaloacetate or α -ketoglutarate; (c) and the addition of ATP. The highest incorporation of acetate into fatty acids was observed in the presence of oxaloacetate (0.02 *M*) and ATP (0.01 *M*) and when the gas phase was air instead of pure O_2 .

We wish to report here the centrifugal fractionation of homogenates of rat mammary gland and which have yielded a soluble enzyme system actively synthesizing fatty acids from [carboxy-¹⁴C]-acetate. The finely minced abdominal glands of lactating rats (7–14 days after parturition) were homogenized with 2 vols of ice-cold buffer (0.154 *M* KCl, 100 pts; 0.154 *M* MgCl₂, 10 pts; and 0.1 *M* potassium phosphate buffer, pH 7.4, 35 pts) in a glass homogenizer provided with a tightly fitting piston which was driven by an electric motor. The homogenate, filtered through four layers of gauze, was centrifuged at 0° and at 400 *g* for 10 min. A pad of fat which separated at the top was scooped off. The supernatant, filtered through a small pad of cotton wool, provided the "full homogenate". It was then centrifuged at 0° and at 25,000 *g* for 30 min. The clear, transparent and pink supernatant (Sp I) was taken off and filtered through a small pad of cotton-wool to remove a thin film of fat from the top. The sediment, designated as "mitochondria", was washed twice by dispersion in fresh buffer and sedimentation at 25,000 *g* for 15 min each. A sample of Sp I was centrifuged further at 2° and 104,000 \times *g* for 30 min at the full speed of the centrifuge. The supernatant (Sp II), which in appearance was similar to Sp I, was taken off and the sediment, a pinkish-brown translucent pellet, designated as "microsomes", was washed once with buffer and sedimented at 104,000 *g* for 10 min at the full speed of the centrifuge. The "mitochondria" obtained from 5 ml of full homogenate were suspended with the aid of a glass homogenizer in 2.5 ml of buffer or with the same volume of Sp I or Sp II to provide 2 ml for one incubation and 0.5 ml for determination of dry weight. The "microsomes" obtained from 10 ml of Sp I were treated in the same way with buffer or Sp II. The incubations were made in WARBURG apparatus at 37.5° for 3 h. Each flask contained 60 μ moles of [carboxy-¹⁴C]-acetate as the sodium salt (5 μ C ¹⁴C), 60 μ moles of potassium oxaloacetate, 30 μ moles of ATP (monosodium salt, L. Light & Co., Ltd. Colnbrook, England) and 2 ml of the preparations to be tested for enzyme activity; the final volume was 3 ml and the gas phase air. In order to obtain sufficient amounts of fatty acids for assay of ¹⁴C, an ethanol-ether solution of the fat obtained during the centrifugal fractionation of the homogenates was added to all preparations, except the full homogenate, at the end of the incubations and before saponification. The fat added was equivalent to 10 mg of mixed fatty acids. The extraction of fatty acids and their purification from contaminating ¹⁴C-acetate by a chromatographic technique has been described elsewhere². The oxygen consumptions ($-Q_{O_2}$) reported represent total O_2 consumed (μ l) per mg dry weight (corrected for salts and fat) of the preparations during the first hr of the experiment, and fatty acid synthesis is expressed as $1 \cdot 10^{-3}$ μ moles acetate used per 100 mg dry weight (corrected) of the preparations.

As can be seen from the results shown in Table I Sp I incorporated 5–10 times as much acetate into fatty acids as the full homogenate or the mitochondria. Moreover, when Sp I was combined with mitochondria fatty acid synthesis was reduced to the lower levels observed with the full homogenate. Sp II, which is a particle-free solution, was less active than Sp I but more active than the full homogenate in synthesizing fatty acids. The recombination of microsomes with Sp II did not restore the enzymic activity to the level observed in Sp I. Whether our inability to disperse the microsomes adequately in Sp II or a loss of some co-factor from the microsomes during washing is the explanation for this result cannot be decided at present. The relatively high O_2 -consumption of Sp I and Sp II indicates the presence of respiratory enzymes. The O_2 -uptake of the preparations cannot be ascribed to autooxidation of components since incubation of (*e.g.*) Sp II without ATP reduced the $-Q_{O_2}$ to 0.9. Spectroscopic analyses of Sp I and II indicated the presence of fairly high concentrations of cytochrome *c* and of DPN.

TABLE I
FATTY ACID SYNTHESIS FROM [*carboxy*-¹⁴C]ACETATE BY FRACTIONS
OF RAT MAMMARY GLAND HOMOGENATES

Preparations incubated	Experiment number							
	112		114		116		117	
	—O ₂	Fatty acid synthesis*	—O ₂	Fatty acid synthesis*	—O ₂	Fatty acid synthesis*	—O ₂	Fatty acid synthesis*
Full homogenate	3.05	6.95			3.06	8.9	2.76	12.1
Sp I	2.81	35.20	2.28	32.4	3.48	90.0	2.92	75.0
			2.41	27.8				
Mitochondria	0.8	6.20	0.99	8.48	—	—	1.09	10.3
			0.84	9.30				
Mitochondria + Sp I	3.19	3.85	3.48	12.90	—	—	2.25	4.54
Sp II			3.27	14.10				
Microsomes	—	—	—	—	3.61	48.5	3.42	18.5
Microsomes	—	—	—	—	0.54	0.0	1.62	0.0
+ Sp II	—	—	—	—	2.95	25.6	2.39	12.4

* Expressed as $1 \cdot 10^{-3}$ μ moles acetate incorporated into fatty acids per 100 mg dry weight.

The mammary gland preparations differ from those made from pigeon-liver (BRADY AND GURIN³) in many respects. To mention only one of these the fractions from pigeon-liver homogenates corresponding to our Sp I and Sp II were inactive and both the particles and supernatant were required for fatty acid synthesis. The inhibition of fatty acid synthesis in Sp I by the mitochondrial fractions could be explained most satisfactorily by competitive reactions.

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Received June 24th, 1953

PREPARATION AND ANALYTICAL DATA OF PURE D(—)3-PHOSPHOGLYCERIC ACID*

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The data in literature concerned with the chemical and physical constants of D(—)3-phosphoglyceric acid are contradictory. Their exact determination seemed desirable in view of the ever increasing significance of this substance for the natural processes of degradation and synthesis.

The acid barium salt of D(—)3-phosphoglyceric acid is readily accessible in crystallized form¹. Its composition is: C₃H₅O₇·PBa·2H₂O. There are, however, wide divergences in the data for crystal

* This investigation was supported by a grant-in-aid from the U.S. Atomic Energy Commission and by a contract between the Office of Naval Research and New York Medical College.